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JB Minireviews—Lipid Signaling

Leukotriene Receptors

Takashi Izumi, Takehiko Yokomizo, Hideru Obinata et al.

Cannabinoid Receptor Ligands

Takayuki Sugiura and Keizo Waku

Nuclear Lipid Signaling

Keiko Tamiya-Koizumi

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- (3) By-line. List full names of all authors. A footnote reference(s) to an author(s), indicating a change of address, should be given on the title-page.
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- (5) Running title. Provide a short running title of less than 60 strokes. It should be as informative as possible.
- (6) The name, complete mailing address, telephone number, and (if accessible) Fax number and E mail address of the person to whom correspondence should be sent. To expedite the review, much of the journal's correspondence will be by Fax, unless the authors request use of regular mail when submitting the manuscript. The Japanese author(s) must also list in Japanese the name and address of the person who is in charge of proofreading.
- (7) Abbreviations. Non-standard abbreviations (see Section

IX-6, 7, and 8) should be defined, even if they are known to those familiar with the field. List all non-standard abbreviations used in the paper in alphabetical order in a footnote on the title page.

2. Summary

- (1) Every paper should have summary. The summary should be concisely written in less than 200 words. Summaries of Rapid Communications should be limited to 100 words. The summary should briefly present the problem, suggest the scope of the work and the plan of experiments, mention significant data and state major findings and conclusions. Avoid statements such as "The significance of these results is discussed" that do not help the reader. The summary should be intelligible to the nonspecialist as well as the specialist in your field, and hence should avoid specialized terms and abbreviations.
- (2) Key words. Provide five key words identifying the nature of the subject matter alphabetically in the last part of the summary.

3. Introduction

The text of a **Regular Paper** should begin with a short introduction with no heading. This should state the reasons for performing the work, with brief reference to previous work on the subject. Avoid giving an extensive review of the literature.

4. Methods, Results, and Discussion

The arrangement of the paper after the introduction is not fixed. The author may separate sections with italicized subheadings.

The Experimental Procedures or Materials and Methods should give sufficient details to enable the reader to repeat your work exactly, if necessary. The necessity for conciseness should not lead to omission of important experimental details. Refer to previously published procedures employed by citation of both the original description and pertinent published modifications, and do not include extensive description unless they present substantially new modifications. Combination of the Results and Discussion in a single section sometimes gives a clearer and more compact presentation.

5. References

References cited in the text should be numbered in parentheses with italicized Arabic numerals in order of appearance. References to "unpublished experiments" and "personal communications" should appear parenthetically in the text following the name(s) of the source of information [(Yamada, T., personal communication), (Suzuki, M. and Yoshida, M., unpublished observations) etc.]. Be sure to verify the wording of any personal communication with the person who supplied the information and get his approval for the use of his name in connection with the quoted information. All references should be listed in numerical order typed double-spaced on a separate sheet under the heading REFERENCES. Please note the following examples.

- (1) For a journal article:
 - Sanger, F., Nicklen, S., and Coulson, A.R. (1977) DNA sequencing with chain-terminating inhibitors. *Proc. Natl.* Acad. Sci. USA 74, 5463-5467
- (2) For a chapter in an edited book:
 - Messing, J. (1983) New M13 vectors for cloning in Methods in Enzymology (Wu, R., Grossman, L., and Moldave, K., eds.) Vol. 101, pp. 20-51, Academic Press, New York
- (3) For a book by one or more authors:
 - Sambrook, J., Fritsch, E.F., and Maniatis, T. (1989) *Molecular Cloning. A Laboratory Manual* pp. 1339-1341, Cold Spring Harbor Laboratory Press, Cold Spring Harbor, NY

Text citations to references written by more than two authors should be styled for example as, Smith *et al.* In the reference list, however, the names of all authors (with initials) must be given. If an article has been accepted for publication by a journal but has not yet appeared in print, the reference should be styled as follows:

 Tanahashi, H. and Ito, T. (1994) Molecular characterization of a novel factor recognizing the interleukin-6 responsive element. J. Biochem. (in press)

The use of "in preparation" and "submitted for publication" is not allowed in the reference list.

Citation of the references written in a language which is usually unreadable for general readers and those published in a journal (or book) to which general reader could not easily access should be avoided.

6. Nucleotide Sequence

New nucleotide data must be submitted and deposited in the DDBJ/EMBL/GenBank databases and an accession number obtained before the paper can be accepted for publication. Submission to any one of the three collaborating databanks is sufficient to ensure data entry in all. The accession number should be included in the manuscript e.g., as a footnote on the title page: "Note: Nucleotide sequence data reported are available in the DDBJ/EMBL/GenBank databases under the accession number(s)....". If requested, the database will withhold release of data until publication. The most convenient method for submitting sequence data is by World Wide Web:

DDBJ via SAKURA: http://sakura.ddbj.nig.ac.jp/

EMBL via WEBIN: http://www.ebi.ac.uk/embl/Submission/

webin. html

GenBank™ via Banklt: http://www.ncbi.nlm.nih.gov/Banklt/ or stand-alone submission tool

Sequin: http://www.ncbi.nlm.nih.gov/Sequin/

For special types of submissions (e.g., genomes, bulk submissions, etc.) additional submission protocols are available from the above sites.

Database Contact Information

DDBJ: Center for Information Biology and DNA Data Bank of Japan National Institute of Genetics, 1111 Yata, Mishima, Shizuoka 411-8540, JAPAN; telephone: +81 559 81 6853; fax: +81 559 81 6849; e-mail: ddbj@ddbj.nig.ac.jp; web URL: http://www.ddbj.nig.ac.jp/

EMBL: EMBL Nucleotide Sequence Submissions, European Bioinformatics Institute, Wellcome Trust Genome Campus, Hinxton, Cambridge DB10 1SD U.K.; telephone: +44 1223 494499; fax: +44 1223 494472; e-mail: datasubs@ebi.ac.uk; web URL: http://www.ebi.ac.uk

GenBank: National Center for Biotechnology Information, National Library of Medicine, Bldg. 38A, Rm 8N-803, Bethesda, MD 20894, U.S.A.; telephone:+1 301 496 2475; fax:+1 301 480 9241; e-mail: info@ncbi.nlm.nih.gov; web URL: http://www.ncbi.nlm.nih.gov

7. Electronic manuscripts

Electronic manuscripts reduce the possibility of introducing errors and resulted in rapid delivery of proofs. After acceptance, authors are encouraged to send the disk plus one printed manuscript to the Editorial Office of the Journal.

V. PREPARATION OF TABLES

- Tables should be drawn on separate sheets and numbered consecutively in Roman numerals. For aid in designing tables in acceptable style, refer to current issues of the Journal.
- Each table should have an explanatory title and sufficient experimental detail, usually in a paragraph immediately following the title, to be intelligible without reference to the text (unless the procedure is given in the Experimental Procedures section, or under another table or figure).
- 3. Indicate units of measure clearly.
- Footnotes to tables should be kept to a minimum and should be indicated by superscript lower cases, at the bottom of the table.

VI. PREPARATION OF ILLUSTRATIONS

- Each figure (Scheme, Diagram) should be given on a separate sheet numbered with an Arabic numeral (Fig. 1, Fig. 2, etc.).
 Figures will be reduced to fit into the type area of the printed page (17.5×23.5 cm).
- 2. Identify all figures in the margin or on the back, with the author's name and figure number and indicate TOP.
- 3. Each figure should be accompanied by a title and an explanatory legend (Legends to Figures). There should be sufficient experimental detail in the legend to make the figure intelligible without reference to the text (unless the same material has been given with a previous figure, or in the Experimental Procedures section).
- Legends to Figures should be typed double-spaced, in numerical order, on a separate page.
- Photographs should be glossy and as high in contrast as possible. Quadruplicate copies for referees should be of the same quality as the original.
- 6. Indicate the magnification of photomicrographs in the legend or

include a bar indicating the scale in the figure.

 Flow diagrams and amino acid or nucleotide sequences should always be presented as direct photographic reproduction.

VII. CHEMICAL AND MATHEMATICAL FORMULAE

- Refer in the text to simple chemical compounds by their formulae when these can be printed in simple horizontal lines of type. Do not use structural formulae in the running text.
- Ionic charge should be shown as a superscript following the chemical symbol, e.g. Fe³⁺, SO₄²⁻.
- 3. Prepare large structural formulae and long mathematical equations in a form suitable for direct photographic reproduction and include them as a Diagram at the end of the paper.
- 4. Isotopically Labeled Compounds—The symbol for an isotope is shown in square brackets directly before the name (word), as in [¹⁴C] urea, [α-¹⁴C] leucine, DL-[methyl-¹⁴C] methionine. When more than one position in a substance is labeled with the same isotope and the positions are not indicated, the number of labeled atoms should be indicated as a right-hand subscript; as in [¹⁴C₂] glycolic acid. The symbol U indicates uniform, e.g. [U-¹⁴C] glycose (where the ¹⁴C is uniformly distributed among all six positions). The isotopic prefix precedes that part of the name to which it refers, as in sodium [¹⁴C] formate, thiamine [β-³²P] diphosphate. Terms such as ¹³¹I-labeled albumin should not be contracted to [¹³¹I] albumin. When isotopes of more than one element are introduced, their symbols should be arranged in alphabetical order: e.g. L-[3-¹⁴C, 2,3-²H, ¹⁵N] serine. The symbols ²H and ³H or D and T may be used for deuterium and tritium, respectively.

For simple molecules, the labeling is indicated by writing the chemical formulae with the prefix superscripts attached to the correct atomic symbols in the formulae: e.g. ¹⁴CO₂, H₂¹⁸O, ²H₂O. Square brackets should not be used for them, or when the isotopic symbol is attached to a word that is not a specific chemical name, abbreviation or symbol: e.g. ¹³¹I-labeled, ¹⁴C-sugar, ¹⁴C-steroids, ³²PO₄³⁻, but [³²P]phosphate.

5. Spectrophotometric Data—Beer's law may be stated as

5. Spectrophotometric Data—Beer's law may be stated as $A = -\log T = \varepsilon lc$

Where A is the absorbance; T, the transmittance $(=I/I_0)$; ε , the molar absorption coefficient; c, the concentration of the absorbing substances in moles per liter; and l, the length of the optical path in centimeters. Under these conditions ε has the dimensions liter•mol⁻¹•cm⁻¹ or more briefly M^{-1} •cm⁻¹ (not cm²•mol⁻¹). Do not use "O.D." and "E."

VIII. ETHICS

In scientific investigations involving human subjects, experiments should be performed in accordance with the ethical standards formulated in the Helsinki Declaration of 1964 (revised in 1989, cf. http://helix.nih.gov:8001/ohsr/helsinki.phtml). Similarly, animal experiments should follow the ethical standards formulated in the Helsinki Declaration, and measures taken to protect animals from pain or discomfort should be mentioned.

IX. TERMINOLOGY AND ABBREVIATIONS

- Abbreviations with specific meanings may be used for convenience for complex chemical substances, particularly in equations, tables, or figures. Avoid using abbreviations in titles and summaries except the standard ones listed in Table II of Section IX-8.
- 2. Use abbreviations and symbols sparingly in the text. In chemical equations, which traditionally depend upon symbols, an abbreviation or symbol may be used for a term that appears in full in the neighboring text. Trivial names are usually sufficiently short not to require abbreviations.
- An abbreviated name or symbol in a column heading in a table, figure, or photograph must either be taken from the "accepted" list given in Section IX-8 or formulated in accordance with the principles of Section IX-6.
- 4. For spelling of chemical names consult current issues of the Journal. For chemical terms follow essentially the usages and rules recommended by International Scientific Union, especially Nomenclature Committee of IUBMB (NC-IUBMB, IUBMB: International Union of Biochemistry and Molecular Biology) and IUPAC-IUBMB Joint Commission on Biochemical Nomencla-

ture (JCBN, IUPAC: International Union of Pure and Applied Chemistry): see the recommendations in Biochemical Nomenclature and Related Documents (1978), available from The Biochemical Society, 7 Warwick Court, London WC1R 5DP, U. K. and in Biochemical Nomenclature and Related Documents. A Compendium, 2nd edn (Liébecq, C., ed.), Portland Press Ltd, London (1992). (see Eur. J. Biochem. 213, 1-3 (1993)).

Refer also to http://www.chem.qmw.ac.uk/iupac/jcbn/

5. Enzymes-Where one or more enzymes figure prominently in a manuscript, authors should use the recommended (trivial) name or systematic name given by Nomenclature Committee of IUBMB and IUPAC-IUBMB Commission on Biochemical Nomenclature: see

Enzyme Nomenclature, Recommendations (1992), Academic Press, Inc.,

see also Eur. J. Biochem. 213, 1-3 (1993).

- Supplement Eur. J. Biochem. 223, 1-5 (1994).
- -Supplement 2 Eur. J. Biochem. 232, 1-6 (1995).
- -Supplement 3 Eur. J. Biochem. 237, 1-5 (1996).
- -Supplement 4 Eur. J. Biochem. 250, 1-6 (1997).

When an enzyme is the main subject of a paper, its source, trivial name, systematic name (or the reaction that it catalyzes) and code number (preceded by "EC") should be included.

- 6. Non-Standard Abbreviations-Use of abbreviations other than the standard ones listed in IX-7 and IX-8 should be kept to a minimum. Such abbreviations should be introduced only when absolutely necessary, as in tables, figures, and other illustrations where space is particularly limited. Abbreviations are usually not needed in the text of a paper where repeated use of long names can be avoided by judicious use of pronouns, or by paraphrasing with words such as "the substrate," "the inhibitor," "the methyl derivative," etc. All non-standard abbreviations used in the text should be defined in alphabetical order in a single footnote on the title page.
- 7. Abbreviations of Units of Measurement and Physical and Chemical Quantities—These abbreviations listed in Table I may be used without definition.

TABLE I

minute

second

(7) Units of Radioactivity

min

1 7 11	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,						
(1)	Prefixes to	o the na	mes of units				
	tera	10^{12}	T	n	nilli	10^{-3}	m
	giga	10^{9}	G	n	nicro	10^{-6}	μ
	mega	10^{6}	M	n	ano	10^{-9}	n
	kilo	10^{3}	k	р	ico	10^{-12}	p
	deci	10-1	deci (not d)	fe	emto	10^{-15}	f
	centi	10^{-2}	$C^{1)}$	a	tto	10^{-18}	а
(2)	Units of (Concentr	ation ²⁾				
	molar (1	moles/li	ter)		M		
	millimo	lar (mill	limoles/liter)		mM (not 10-	3 M)
	microme	olar (mi	cromoles/lite	r)	μ M (or 10 ⁻⁶ l	M)
	nanomo	lar (nan	omoles/liter)			or ×10-	
	picomol	ar (picoı	noles/liter)		pM (c	$r \times 10^{-}$	12 M)
(3)	Units of I	_ength					
	meter				m		
	centime	ter			cm		
	millime	ter			mm		
	microme	eter (not	t micron)		μm (1	$\cot \mu$)	
	nanome				nm (r	not mμ)	
	Ångstro				Å		
(4)	Units of A						
	square o	centimet	er		cm^2		
	cubic ce	ntimete	r		cm^3		
	liter				l (in t	tables or	ıly)
	millilite	r			ml		
	microlit	er			μ l (no	otλ)	
(5)	Units of M	Mass .					
	gram				g (kg,	mg, μ g	[not γ],
					ng,	pg)	
	dalton³)				Da		
(6)	Units of T						
	hour	ŀ	1	year	r	yr	

month

week

day

mo

wk

d

becquerel	Bq (=1 dps or 60 dpm)
counts per minute	cpm
curie(s)	Ci $(=3.7\times10^{10} \text{ Bq})$
disintegrations per minute	dpm
(8) Other Units	•
mole	mol (mmol, μmol, nmol, pmol)
degree Celsius	°C
degree absolute (kelvin)	K
joule	J
kilojoule	kJ
calorie	cal
kilocalorie	kcal
parts per billion	ppb
parts per million	ppm
cycles per second (hertz)	Hz (not cps)
equivalent	eq
ampere	A (mA)
ohm	Ω
volt	V
	v G
gauss pascal	Pa
•	
revolutions per minute	rpm
Svedberg unit of sedimentation coefficient (10 ⁻¹³ s)	S
	b
(9) Physical and Chemical Quantities	4
absorbance	A K
equilibrium constant	k
rate constant	
maximum velocity	V_{\max}
Michaelis constant	K_{m}
equilibrium dissociation con-	V
stant	$K_{ m d}$ pI
isoelectric point	M_r
molecular weight ³⁾	•
retardation factor	R_f
acceleration of gravity	g
specific rotation	$[\alpha]_{\lambda}^{t}$
partial specific volume	<u>v</u>
diffusion constant	D
sedimentation coefficient	s
density	ρ
sedimentation coefficient in	
water	
at 20°C, extraporated to zero	•
concentration	$s_{20,\mathbf{w}}^0$
Gibbs energy change	ΔG
entropy change	∆S
enthalpy change	∆H
melting temperature	$T_{ m m}$
(10) Other Terms	
logarithm	log
logarithm (natural)	ln
standard deviation of a series	SD
standard error of mean of series	SE
0	

1) to be avoided where possible (except for cm).

2) Terms such as milligram percent (mg%) should not be used. Weight concentrations should be given as g/ml, g/100 ml, etc. 3) Molecular weight is dimensionless. Only molecular mass is expressed by daltons.

8. Accepted Abbreviations and Symbols-Authors may use, without definition, the abbreviations given in Table II and the symbols and abbreviations for amino acid or nucleotide residues in polymers or sequences. Define other abbreviations in a single footnote on the title page.

TABLE II

(1) (1	
(1) General	
Adenosine 3':5'-cyclic monophosphate	cAMP
Adenosine 5'-mono-, di, and triphos-	AMP, ADP, and
phates ¹⁾	ATP
Adenosine triphosphatase	ATPase

_	
Base pair(s)	bp
Bovine serum albumin O-(Carboxymethyl)	BSA CM-
Circular dichroism	CD
Coenzyme A and its acyl derivatives	CoA (or CoASH)
•	and acyl-CoA
Complementary DNA	cDNA
Cyclic AMP	cAMP
Cyclic GMP	cGMP
Cytidine diphosphate choline, etc. Cytidine 5'-mono-, di-, and triphos-	CDP-choline, etc. CMP, CDP, and
phates	CTP
Deoxyribonuclease	DNase
Deoxyribonucleic acid	DNA
O-(Diethylaminoethyl)	DEAE-
Dithiothreitol	DTT
Electron paramagnetic resonance	EPR ESR
Electron spin resonance Ethylenediaminetetraacetic acid	EDTA
[Ethylenebis(oxyethlenenitrilo)]-	EGTA
tetraacetic acid	20111
Flavin-adenine dinucleotide and its	
fully reduced form	FAD and FADH ₂
Flavin mononucleotide and its fully	
reduced form	FMN and FMNH ₂
Fourier transform	FT
Gas chromatography-mass spectrom- etry	GC-MS
Gas liquid chromatography	GLC
Glutathione and its oxidized form	GSH and GSSG
Guanosine 3':5'-cyclic monophosphate	cGMP
Guanosine 5'-mono-, di-, and triphos-	GMP, GDP, and
phates	GTP
Guanosine triphosphatase	GTPase
Hemoglobin	Hb hnRNA
Heterogenous nuclear RNA High performance (pressure) liquid	IIIIKINA
chromatography	HPLC
4-(2-Hydroxyethyl)-1-piperazineethane-	
sulfonic acid	HEPES
Immunoglobulin	Ig (IgG, IgM, etc.)
Infrared	IR
Inorganic orthophosphate	P _i
Inorganic pyrophosphate Inosine 5'-mono-, di-, and triphos-	PP ₁ IMP, IDP, and ITP
phates	IMI, IDI, and III
Kilobases	kb
Kilobase pairs	kbp
Lethal dose, 50%	LD_{50}
Messenger RNA	mRNA
Nicotinamide adenine dinucleotide and	NIAD+ 1 NIADIT?\
its reduced form	NAD ⁺ and NADH ²⁾ NADP ⁺ and
Nicotinamide adenine dinucleotide phosphate and its reduced form	NADPH ²⁾
Nuclear magnetic resonance	NMR
Nuclear RNA	nRNA
Optical rotatory dispersion	ORD
Phosphoric acid residue	P- or -P
Pseudouridine and pseudouridine	
mono-	d and dMD
nucleotide Polyacrylamide gel electrophoresis	ψ and ψ MP PAGE
Poly(adenylic acid), polyadenylate ³⁾	Poly(A)3)
Polymerase chain reaction	PCR
Restriction fragment length polymor-	RFLP
phism	
Ribonuclease	RNase
Ribonucleic acid	RNA
Ribosylthymine 5' mano, dia and tri-	rRNA TMP, TDP, and
Ribosylthymine 5'-mono-, di-, and tri- phosphates	TMP, TDP, and
Sodium dodecyl sulfate	* * *
	SDS
Thin layer chromatography	SDS TLC
Thymidine (2'-deoxyribosylthymine)	TLC dTMP, dTDP, and
	TLC

Uridine 5'-mono-, di-, and triphosphates (2) Amino acids Alanine Arg (R) Asparagine Aspartic acid Glu (E) Glutamic acid Glu (E) Gly (G) Histidine Glx (Z) Glycine Gly (G) Histidine His (H) Isoleucine Ile (I) Leucine Leu (L) Lysine Methionine Met (M) Phenylalanine Phe (F) Pro (P) Serine Ser (S) Threonine Thr (T) Tryptophan Tryrosine Yaline (3) Nucleic acids Adenosine Bromouridine Cytidine Obihydrouridine Guanosine Inosine G-Mercaptopurine ribonucleoside (6-thioinosine) 'a nucleoside' 'a purine nucleoside' 'a pyrimidine nucleoside' 'a pyrimidine nucleoside' 'a pyrimidine (2'-deoxyribosylthymine) Uridine Xanthosine Po or p The various isomers of adenosine monophosphate may	Transfer RNA Tris(hydroxymethyl)aminomethane Ultraviolet Uridine diphosphate glucose, etc Uridine 5'-mono-, di-, and triphos-	tRNA Tris UV UDP-gl	icose, <i>etc</i> .
(2) Amino acids Alanine Ala (A) Arginine Asparagine Asparatic acid Aspartic acid or asparagine Cysteine Cystein			,
Alanine Arginine Arginine Asparagine Asparagine Aspartic acid Aspartic acid or asparagine Cysteine Cysteine Cysteine Cysteine Cys (C) Glutamic acid Glu (E) Glutamine Glu (E) Glutamine Glu (E) Glutamic acid or glutamine Glx (Z) Glycine Gly (G) Histidine His (H) Isoleucine Ile (I) Leucine Leu (L) Lysine Methionine Met (M) Phenylalanine Phe (F) Proline Serine Ser (S) Threonine Thr (T) Tryptophan Trp (W) Tyrosine Tyr (Y) Valine (3) Nucleic acids Adenosine Bromouridine G-Mercaptopurine ribonucleoside (6-thioinosine) 'a nucleoside' Apurine nucleoside' 'a pyrimidine nucleoside' 'a pyrimidine nucleoside' 'a pyrimidine Thymidine (2'-deoxyribosylthymine) Uridine Xanthosine Res (N) Aspar (R) Aspar (N) Aspar			
Arginine Asparagine Asparatic acid Aspartic acid or asparagine Asy (B) Aspartic acid or asparagine Asy (B) Cysteine Cysteine Cysteine Cysteine Glutamic acid Glu (E) Glutamic acid or glutamine Glx (Z) Glycine Gly (G) Histidine His (H) Isoleucine Leu (L) Lysine Methionine Met (M) Phenylalanine Phe (F) Proline Pro (P) Serine Ser (S) Threonine Thr (T) Tryptophan Trp (W) Tyrosine Tyr (Y) Valine Val (V) (3) Nucleic acids Adenosine Bromouridine Bromouridine Guanosine G-Mercaptopurine ribonucleoside (G-thioinosine) 'a nucleoside' 'a purine nucleoside' 'a pyrimidine nucleoside' 'a pyrimidine nucleoside' Thymidine (2'-deoxyribosylthymine) Uridine Xanthosine Po (C) Glutamic Asp (D) Aspartic Aspa (D) Aspa (D) Aspa (D) Aspa		Ala	(A)
Asparagine		Arg	
Aspartic acid or asparagine			
Aspartic acid or asparagine Cysteine Cysteine Cysteine Clutamic acid Glu Glutamine Gln Gln Gly Gly Gly Gly Gly Gly Histidine His Isoleucine Ile Ile Il Leucine Leu Lys Ky Methionine Met Met My Phenylalanine Phe Fo Proline Serine Ser Ser So Threonine Thr Tryptophan Try Wy Tyrosine Tyr Valine Val (V) (3) Nucleic acids Adenosine Bromouridine Cytidine Cytidine Inosine G-Mercaptopurine ribonucleoside (6-thioinosine) 'a nucleoside' Pseudouridine Thymidine (2'-deoxyribosylthymine) Uridine Xanthosine X Phosphoric residue Asx (B) Cys (C) Glu (B) Glu (B) Glu (B) Gly (G) His (G) His (I) Leu (I)			
Cysteine Glutamic acid Glutamine Glu Glutamine Glu Glutamic acid or glutamine Glx Gly Gly Gly Gly Histidine His Isoleucine Ile Ile Ile Ile Ile Leucine Leu Lys K Methionine Met Met MM Phenylalanine Phe Proline Pro Serine Ser Sor Threonine Thr Tryptophan Tyrytophan Tyrytophan Tyryosine Val Ive I(V) Inc		_	1
Glutamic acid Glu (E) Glutamine Gln (Q) Glutamic acid or glutamine Glx (Z) Glycine Gly (G) Histidine His (H) Isoleucine Ile (I) Leucine Leu (L) Lysine Lys (K) Methionine Met (M) Phenylalanine Phe (F) Proline Pro (P) Serine Ser (S) Threonine Thr (T) Tryptophan Trp (W) Tyrosine Tyr (Y) Valine Val (V) (3) Nucleic acids A BrUrd or B Cytidine D or hU G Cytidine D or hU G Guanosine G M or sI (6-thioinosine) Nuc or N 'a nucleoside' Nuc or N 'a pyrimidine nucleoside' Y<			1 . 1
Glutamine Gln Q) Glutamic acid or glutamine Glx (Z) Glycine Gly (G) Histidine His (H) Isoleucine Ile (I) Leucine Leu (L) Lysine Lys (K) Methionine Met (M) Phenylalanine Phe (F) Proline Pro (P) Serine Ser (S) Threonine Thr (T) Tryptophan Trp (W) Tyrosine Tyr (Y) Valine Val (V) (3) Nucleic acids A BrUrd or B Adenosine A BrUrd or B Cytidine D or hU Guanosine G Inosine I G 6-Mercaptopurine ribonucleoside M or sI (6-thioinosine) M or sI 'a purine nucleoside' Y Thouridine S or sU <t< td=""><td> T</td><td>•</td><td>1.1</td></t<>	T	•	1.1
Glutamic acid or glutamine Glx (Z) Glycine Gly (G) Histidine His (H) Isoleucine Ile (I) Leucine Leu (L) Lysine Lys (K) Methionine Met (M) Phenylalanine Phe (F) Proline Pro (P) Serine Ser (S) Threonine Thr (T) Tryptophan Trp (W) Tyrosine Tyr (Y) Valine Val (V) (3) Nucleic acids A BrUrd or B Cytidine D or hU BrUrd or B Cytidine D or hU Guanosine G Inosine I M or sI 6-Mercaptopurine ribonucleoside M or sI W or Qa 'a purine nucleoside' Nuc or N Thouridine S or sU Thymidine (2'-deoxyribosylthymine) U <td< td=""><td></td><td></td><td></td></td<>			
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1) The various isomers of adenosine monophosphate may be written 2'-AMP, 3'-AMP, or 5'-AMP (in case of possible ambiguity). A similar procedure may be a similar procedure may be a similar procedure. ambiguity). A similar procedure may be applied to other nucleoside or deoxyribonucleoside monophosphates.

²⁾ NAD(P)⁺ and NAD(P)H indicate either NAD⁺ or NADP⁺ and either NADH or NADPH, respectively.

3) Similarly abbreviate oligo- and polynucleotides composed of repeating sequences or of unknown sequence of given purine or pyrimidine bases, e.g. oligothymidylate, oligo(dT); alternating copolymer of A and U, poly(A-U); random copolymer of A and U, poly(A,U).

The d prefix may be used to represent the corresponding

deoxyribonucleoside phosphates, e.g. dADP.

- 9. Names of Animals, Plants, and Microorganisms-The scientific names are Latin binomials and should be given in full in the title and summary and on first mention in the text (e.g. Escherichia coli). Subsequently, the generic name may be contracted (usually to the first letter), e.g., E. coli. The strain of laboratory animals and if possible the source should be stated.
- 10. The cytochromes should be designated by a small italicized letter, e.g. cytochrome a, b_2 , c_1 , etc.

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